EXHIBIT A
(PART 3 OF 4)

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accept the electrochemical sensor and the inserter. The insertion gun has a driving mechanism for driving the inserter and electrochemical sensor into the patient, and a retraction mechanism for removing the inserter while leaving the sensor within the patient.

Another embodiment is a method of using an electrochemical sensor. A mounting unit is adhered to skin of a patient. An insertion gun is aligned with a port on the mounting unit. The electrochemical sensor is disposed within the insertion gun and then the electrochemical sensor is inserted into the skin of the patient using the insertion gun. The insertion gun is removed and a housing of the sensor control unit is mounted on the mounting base. A plurality of conductive contacts disposed on the housing is coupled to a plurality of contact pads disposed on the electrochemical sensor to prepare the sensor for use.

One embodiment of the invention is a method for detecting failures in an implanted analyte-responsive sensor. An analyte-responsive sensor is implanted into a patient. The analyte-responsive sensor includes N working electrodes, where N is an integer and is two or greater, and a common counter electrode. Signals generated at one of the N working electrodes and at the common counter electrode are then obtained and the sensor is determined to have failed if the signal from the common counter electrode is not N times the signal from one of the working electrodes, within a predetermined threshold limit.

Yet another embodiment is a method of calibrating an electrochemical sensor having one or more working electrodes implanted in a patient. A signal is generated from each of the working electrodes. Several conditions are tested to determine if calibration is appropriate. First, the signals from each of the one or more working electrodes should differ by less than a first threshold amount. Second, the signals from each of the one or more working electrodes should be within a predetermined range. And, third, a rate of change of the signals from each of the one or more working electrodes should be less than a second threshold amount. A calibration value is found assaying a calibration sample of a patient's body fluid. The calibration value is then related to at least one of the signals from the one or more working electrodes if the conditions described above are met.

A further embodiment is a method for monitoring a level of an analyte. A sensor is inserted into a skin of a patient and a sensor control unit is attached to the skin of the patient. Two or more conductive contacts on the sensor control unit are coupled to contact pads on the sensor. Then, using the sensor control unit, data is collected regarding a level of an analyte from signals generated by the sensor. The collected data is transmitted to a display unit and an indication of the level of the analyte is displayed on the display unit.

The above summary of the present invention is not intended to describe each disclosed embodiment or every implementation of the present invention. The Figures and the detailed description which follow more particularly exemplify these embodiments.

BRIEF DESCRIPTION OF THE DRAWINGS

The invention may be more completely understood in consideration of the following detailed description of various embodiments of the invention in connection with the accompanying drawings, in which:

FIGS. 1, 1A, 1B is a block diagram of one embodiment of a subcutaneous analyte monitor using a subcutaneously implantable analyte sensor, according to the invention;

FIG. 2 is a top view of one embodiment of an analyte sensor, according to the invention;

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FIGS. 3, 3A is a cross-sectional view of the analyte sensor of FIG. 2;

FIG. 3B is a cross-sectional view of another embodiment of an analyte sensor, according to the invention;

FIG. 4A is a cross-sectional view of a third embodiment of an analyte sensor, according to the invention;

FIG. 4B is a cross-sectional view of a fourth embodiment of an analyte sensor, according to the invention;

FIGS. 5A–5D is an expanded top view of a tip portion of the analyte sensor of FIG. 2;

FIG. 6 is a cross-sectional view of a fifth embodiment of an analyte sensor, according to the invention;

FIG. 7 is an expanded top view of a tip-portion of the analyte sensor of FIG. 6;

FIG. 8 is an expanded bottom view of a tip-portion of the analyte sensor of FIG. 6;

FIG. 9 is a side view of the analyte sensor of FIG. 2;

FIG. 10 is a top view of the analyte sensor of FIG. 6;

FIG. 11 is a bottom view of the analyte sensor of FIG. 6;

FIG. 12 is an expanded side view of one embodiment of a sensor and an insertion device, according to the invention;

FIGS. 13A, 13B, 13C are cross-sectional views of three embodiments of the insertion device of FIG. 12;

FIG. 14 is a cross-sectional view of one embodiment of a on-skin sensor control unit, according to the invention;

FIG. 15 is a top view of a base of the on-skin sensor control unit of FIG. 14;

FIG. 16 is a bottom view of a cover of the on-skin sensor control unit of FIG. 14;

FIG. 17 is a perspective view of the on-skin sensor control unit of FIG. 14 on the skin of a patient;

FIG. 18A is a block diagram of one embodiment of an on-skin sensor control unit, according to the invention;

FIG. 18B is a block diagram of another embodiment of an on-skin sensor control unit, according to the invention;

FIGS. 19A, 19B, 19C, and 19D are cross-sectional views of four embodiments of conductive contacts disposed on an interior surface of a housing of an on-skin sensor control unit, according to the invention;

FIGS. 19E and 19F are cross-sectional views of two embodiments of conductive contacts disposed on an exterior surface of a housing of an on-skin sensor control unit, according to the invention;

FIGS. 20A and 20B are schematic diagrams of two embodiments of a current-to-voltage converter for use in an analyte monitoring device, according to the invention;

FIG. 21 is a block diagram of one embodiment of an open loop modulation system for use in an analyte monitoring device, according to the invention;

FIG. 22 is a block diagram of one embodiment of a receiver/display unit, according to the invention;

FIG. 23 is a front view of one embodiment of a receiver/display unit;

FIG. 24 is a front view of a second embodiment of a receiver/display unit;

FIG. 25 is a block diagram of one embodiment of a drug delivery system, according to the invention;

FIG. 26 is a perspective view of the internal structure of an insertion gun, according to the invention;

FIG. 27A is a top view of one embodiment of an on-skin sensor control unit, according to the invention;

FIG. 27B is a top view of one embodiment of a mounting unit of the on-skin sensor control unit of FIG. 27A;

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FIG. **28**A is a top view of another embodiment of an on-skin sensor control unit after insertion of an insertion device and a sensor, according to the invention;

FIG. **28**B is a top view of one embodiment of a mounting unit of the on-skin sensor control unit of FIG. **28**A;

FIG. **28**C is a top view of one embodiment of a housing for at least a portion of the electronics of the on-skin sensor control unit of FIG. **28**A;

FIG. **28**D is a bottom view of the housing of FIG. **28**C; and

FIG. **28**E is a top view of the on-skin sensor control unit of FIG. **28**A with a cover of the housing removed.

While the invention is amenable to various modifications and alternative forms, specifics thereof have been shown by way of example in the drawings and will be described in detail. It should be understood, however, that the intention is not to limit the invention to the particular embodiments described. On the contrary, the intention is to cover all modifications, equivalents, and alternatives falling within the spirit and scope of the invention as defined by the appended claims.

DETAILED DESCRIPTION OF THE INVENTION

The present invention is applicable to an analyte monitoring system using an implantable sensor for the in vivo determination of a concentration of an analyte, such as glucose or lactate, in a fluid. The sensor can be, for example, subcutaneously implanted in a patient for the continuous or periodic monitoring an analyte in a patient's interstitial fluid. This can then be used to infer the glucose level in the patient's bloodstream. Other in vivo analyte sensors can be made, according to the invention, for insertion into a vein, artery, or other portion of the body containing fluid. The analyte monitoring system is typically configured for monitoring the level of the analyte over a time period which may range from days to weeks or longer.

The following definitions are provided for terms used herein:

A "counter electrode" refers to an electrode paired with the working electrode, through which passes a current equal in magnitude and opposite in sign to the current passing through the working electrode. In the context of the invention, the term "counter electrode" is meant to include counter electrodes which also function as reference electrodes (i.e., a counter/reference electrode).

An "electrochemical sensor" is a device configured to detect the presence and/or measure the level of an analyte in a sample via electrochemical oxidation and reduction reactions on the sensor. These reactions are transduced to an electrical signal that can be correlated to an amount, concentration, or level of an analyte in the sample.

"Electrolysis" is the electrooxidation or electroreduction of a compound either directly at an electrode or via one or more electron transfer agents.

A compound is "immobilized" on a surface when it is entrapped on or chemically bound to the surface.

A "non-leachable" or "non-releasable" compound or a compound that is "non-leachably disposed" is meant to define a compound that is affixed on the sensor such that it does not substantially diffuse away from the working surface of the working electrode for the period in which the sensor is used (e.g., the period in which the sensor is implanted in a patient or measuring a sample).

Components are "immobilized" within a sensor, for example, when the components are covalently, ionically, or

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coordinatively bound to constituents of the sensor and/or are entrapped in a polymeric or sol-gel matrix or membrane which precludes mobility.

An "electron transfer agent" is a compound that carries electrons between the analyte and the working electrode, either directly, or in cooperation with other electron transfer agents. One example of an electron transfer agent is a redox mediator.

A "working electrode" is an electrode at which the analyte (or a second compound whose level depends on the level of the analyte) is electrooxidized or electroreduced with or without the agency of an electron transfer agent.

A "working surface" is that portion of the working electrode which is coated with or is accessible to the electron transfer agent and configured for exposure to an analyte-containing fluid.

A "sensing layer" is a component of the sensor which includes constituents that facilitate the electrolysis of the analyte. The sensing layer may include constituents such as an electron transfer agent, a catalyst which catalyzes a reaction of the analyte to produce a response at the electrode, or both. In some embodiments of the sensor, the sensing layer is non-leachably disposed in proximity to or on the working electrode.

A "non-corroding" conductive material includes non-metallic materials, such as carbon and conductive polymers.

Analyte Sensor Systems

The analyte monitoring systems of the present invention can be utilized under a variety of conditions. The particular configuration of a sensor and other units used in the analyte monitoring system may depend on the use for which the analyte monitoring system is intended and the conditions under which the analyte monitoring system will operate. One embodiment of the analyte monitoring system includes a sensor configured for implantation into a patient or user. For example, implantation of the sensor may be made in the arterial or venous systems for direct testing of analyte levels in blood. Alternatively, a sensor may be implanted in the interstitial tissue for determining the analyte level in interstitial fluid. This level may be correlated and/or converted to analyte levels in blood or other fluids. The site and depth of implantation may affect the particular shape, components, and configuration of the sensor. Subcutaneous implantation may be preferred, in some cases, to limit the depth of implantation of the sensor. Sensors may also be implanted in other regions of the body to determine analyte levels in other fluids. Examples of suitable sensor for use in the analyte monitoring systems of the invention are described in U.S. patent application, Ser. No. 09/034,372, incorporated herein by reference.

One embodiment of the analyte monitoring system **40** for use with an implantable sensor **42**, and particularly for use with a subcutaneously implantable sensor, is illustrated in block diagram form in FIG. **1**. The analyte monitoring system **40** includes, at minimum, a sensor **42**, a portion of which is configured for implantation (e.g., subcutaneous, venous, or arterial implantation) into a patient, and a sensor control unit **44**. The sensor **42** is coupled to the sensor control unit **44** which is typically attached to the skin of a patient. The sensor control unit **44** operates the sensor **42**, including, for example, providing a voltage across the electrodes of the sensor **42** and collecting signals from the sensor **42**. The sensor control unit **44** may evaluate the signals from the sensor **42** and/or transmit the signals to one or more optional receiver/display units **46, 48** for evaluation. The sensor control unit **44** and/or the receiver/display units

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46, 48 may display or otherwise communicate the current level of the analyte. Furthermore, the sensor control unit **44** and/or the receiver/display units **46, 48** may indicate to the patient, via, for example, an audible, visual, or other sensory-stimulating alarm, when the level of the analyte is at or near a threshold level. In some embodiments, a electrical shock can be delivered to the patient as a warning through one of the electrodes or the optional temperature probe of the sensor. For example, if glucose is monitored then an alarm may be used to alert the patient to a hypoglycemic or hyperglycemic glucose level and/or to impending hypoglycemia or hyperglycemia.

The Sensor

A sensor **42** includes at least one working electrode **58** formed on a substrate **50**, as shown in FIG. **2**. The sensor **42** may also include at least one counter electrode **60** (or counter/reference electrode) and/or at least one reference electrode **62** (see FIG. **8**). The counter electrode **60** and/or reference electrode **62** may be formed on the substrate **50** or may be separate units. For example, the counter electrode and/or reference electrode may be formed on a second substrate which is also implanted in the patient or, for some embodiments of the implantable sensors, the counter electrode and/or reference electrode may be placed on the skin of the patient with the working electrode or electrodes being implanted into the patient. The use of an on-the-skin counter and/or reference electrode with an implantable working electrode is described in U.S. Pat. No. 5,593, 852, incorporated herein by reference.

The working electrode or electrodes **58** are formed using conductive traces **52** disposed on the substrate **50**. The counter electrode **60** and/or reference electrode **62**, as well as other optional portions of the sensor **42**, such as a temperature probe **66** (see FIG. **8**), may also be formed using conductive traces **52** disposed on the substrate **50**. These conductive traces **52** may be formed over a smooth surface of the substrate **50** or within channels **54** formed by, for example, embossing, indenting or otherwise creating a depression in the substrate **50**.

A sensing layer **64** (see FIGS. **3A** and **3B**) is often formed proximate to or on at least one of the working electrodes **58** to facilitate the electrochemical detection of the analyte and the determination of its level in the sample fluid, particularly if the analyte can not be electrolyzed at a desired rate and/or with a desired specificity on a bare electrode. The sensing layer **64** may include an electron transfer agent to transfer electrons directly or indirectly between the analyte and the working electrode **58**. The sensing layer **64** may also contain a catalyst to catalyze a reaction of the analyte. The components of the sensing layer may be in a fluid or gel that is proximate to or in contact with the working electrode **58**. Alternatively, the components of the sensing layer **64** may be disposed in a polymeric or sol-gel matrix that is proximate to or on the working electrode **58**. Preferably, the components of the sensing layer **64** are non-leachably disposed within the sensor **42**. More preferably, the components of the sensor **42** are immobilized within the sensor **42**.

In addition to the electrodes **58, 60, 62** and the sensing layer **64**, the sensor **42** may also include a temperature probe **66** (see FIGS. **6** and **8**), a mass transport limiting layer **74** (see FIG. **9**), a biocompatible layer **75** (see FIG. **9**), and/or other optional components, as described below. Each of these items enhances the functioning of and/or results from the sensor **42**, as discussed below.

The Substrate

The substrate **50** may be formed using a variety of non-conducting materials, including, for example, poly-

meric or plastic materials and ceramic materials. Suitable materials for a particular sensor **42** may be determined, at least in part, based on the desired use of the sensor **42** and properties of the materials.

In some embodiments, the substrate is flexible. For example, if the sensor **42** is configured for implantation into a patient, then the sensor **42** may be made flexible (although rigid sensors may also be used for implantable sensors) to reduce pain to the patient and damage to the tissue caused by the implantation of and/or the wearing of the sensor **42**. A flexible substrate **50** often increases the patient's comfort and allows a wider range of activities. Suitable materials for a flexible substrate **50** include, for example, non-conducting plastic or polymeric materials and other non-conducting, flexible, deformable materials. Examples of useful plastic or polymeric materials include thermoplastics such as polycarbonates, polyesters (e.g., Mylar™ and polyethylene terephthalate (PET)), polyvinyl chloride (PVC), polyurethanes, polyethers, polyamides, polyimides, or copolymers of these thermoplastics, such as PETG (glycol-modified polyethylene terephthalate).

In other embodiments, the sensors **42** are made using a relatively rigid substrate **50** to, for example, provide structural support against bending or breaking. Examples of rigid materials that may be used as the substrate **50** include poorly conducting ceramics, such as aluminum oxide and silicon dioxide. One advantage of an implantable sensor **42** having a rigid substrate is that the sensor **42** may have a sharp point and/or a sharp edge to aid in implantation of a sensor **42** without an additional insertion device.

It will be appreciated that for many sensors **42** and sensor applications, both rigid and flexible sensors will operate adequately. The flexibility of the sensor **42** may also be controlled and varied along a continuum by changing, for example, the composition and/or thickness of the substrate **50**.

In addition to considerations regarding flexibility, it is often desirable that implantable sensors **42** should have a substrate **50** which is non-toxic. Preferably, the substrate **50** is approved by one or more appropriate governmental agencies or private groups for in vivo use.

The sensor **42** may include optional features to facilitate insertion of an implantable sensor **42**, as shown in FIG. **12**. For example, the sensor **42** may be pointed at the tip **123** to ease insertion. In addition, the sensor **42** may include a barb **125** which assists in anchoring the sensor **42** within the tissue of the patient during operation of the sensor **42**. However, the barb **125** is typically small enough that little damage is caused to the subcutaneous tissue when the sensor **42** is removed for replacement.

Although the substrate **50** in at least some embodiments has uniform dimensions along the entire length of the sensor **42**, in other embodiments, the substrate **50** has a distal end **67** and a proximal end **65** with different widths **53, 55**, respectively, as illustrated in FIG. **2**. In these embodiments, the distal end **67** of the substrate **50** may have a relatively narrow width **53**. For sensors **42** which are implantable into the subcutaneous tissue or another portion of a patient's body, the narrow width **53** of the distal end **67** of the substrate **50** may facilitate the implantation of the sensor **42**. Often, the narrower the width of the sensor **42**, the less pain the patient will feel during implantation of the sensor and afterwards.

For subcutaneously implantable sensors **42** which are designed for continuous or periodic monitoring of the analyte during normal activities of the patient, a distal end **67** of the sensor **42** which is to be implanted into the patient has

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a width 53 of 2 mm or less, preferably 1 mm or less, and more preferably 0.5 mm or less. If the sensor 42 does not have regions of different widths, then the sensor 42 will typically have an overall width of, for example, 2 mm, 1.5 mm, 1 mm, 0.5 mm, 0.25 mm, or less. However, wider or narrower sensors may be used. In particular, wider implantable sensors may be used for insertion into veins or arteries or when the movement of the patient is limited, for example, when the patient is confined in bed or in a hospital.

Returning to FIG. 2, the proximal end 65 of the sensor 42 may have a width 55 larger than the distal end 67 to facilitate the connection between contact pads 49 of the electrodes and contacts on a control unit. The wider the sensor 42 at this point, the larger the contact pads 49 can be made. This may reduce the precision needed to properly connect the sensor 42 to contacts on the control unit (e.g., sensor control unit 44 of FIG. 1). However, the maximum width of the sensor 42 may be constrained so that the sensor 42 remains small for the convenience and comfort of the patient and/or to fit the desired size of the analyte monitor. For example, the proximal end 65 of a subcutaneously implantable sensor 42, such as the sensor 42 illustrated in FIG. 1, may have a width 55 ranging from 0.5 mm to 15 mm, preferably from 1 mm to 10 mm, and more preferably from 3 mm to 7 mm. However, wider or narrower sensors may be used in this and other in vivo applications.

The thickness of the substrate 50 may be determined by the mechanical properties of the substrate material (e.g., the strength, modulus, and/or flexibility of the material), the desired use of the sensor 42 including stresses on the substrate 50 arising from that use, as well as the depth of any channels or indentations formed in the substrate 50, as discussed below. Typically, the substrate 50 of a subcutaneously implantable sensor 42 for continuous or periodic monitoring of the level of an analyte while the patient engages in normal activities has a thickness of 50 to 500 μm and preferably 100 to 300 μm. However, thicker and thinner substrates 50 may be used, particularly in other types of in vivo sensors 42.

The length of the sensor 42 may have a wide range of values depending on a variety of factors. Factors which influence the length of an implantable sensor 42 may include the depth of implantation into the patient and the ability of the patient to manipulate a small flexible sensor 42 and make connections between the sensor 42 and the sensor control unit 44. A subcutaneously implantable sensor 42 for the analyte monitor illustrated in FIG. 1 may have a length ranging from 0.3 to 5 cm, however, longer or shorter sensors may be used. The length of the narrow portion of the sensor 42 (e.g., the portion which is subcutaneously inserted into the patient), if the sensor 42 has narrow and wide portions, is typically about 0.25 to 2 cm in length. However, longer and shorter portions may be used. All or only a part of this narrow portion may be subcutaneously implanted into the patient. The lengths of other implantable sensors 42 will vary depending, at least in part, on the portion of the patient into which the sensor 42 is to be implanted or inserted.

Conductive Traces

At least one conductive trace 52 is formed on the substrate for use in constructing a working electrode 58. In addition, other conductive traces 52 may be formed on the substrate 50 for use as electrodes (e.g., additional working electrodes, as well as counter, counter/reference, and/or reference electrodes) and other components, such as a temperature probe. The conductive traces 52 may extend most of the distance along a length 57 of the sensor 50, as illustrated in FIG. 2, although this is not necessary. The placement of the conductive traces 52 may depend on the particular configuration of the analyte monitoring system (e.g., the placement of control unit contacts and/or the sample chamber in relation to the sensor 42). For implantable sensors, particularly subcutaneously implantable sensors, the conductive traces typically extend close to the tip of the sensor 42 to minimize the amount of the sensor that must be implanted.

The conductive traces 52 may be formed on the substrate 50 by a variety of techniques, including, for example, photolithography, screen printing, or other impact or non-impact printing techniques. The conductive traces 52 may also be formed by carbonizing conductive traces 52 in an organic (e.g., polymeric or plastic) substrate 50 using a laser. A description of some exemplary methods for forming the sensor 42 is provided in U.S. patent application Ser. No. 09/034,422, incorporated herein by reference.

Another method for disposing the conductive traces 52 on the substrate 50 includes the formation of recessed channels 54 in one or more surfaces of the substrate 50 and the subsequent filling of these recessed channels 54 with a conductive material 56, as shown in FIG. 3A. The recessed channels 54 may be formed by indenting, embossing, or otherwise creating a depression in the surface of the substrate 50. Exemplary methods for forming channels and electrodes in a surface of a substrate can be found in U.S. patent application Ser. No. 09/034,422. The depth of the channels is typically related to the thickness of the substrate 50. In one embodiment, the channels have depths in the range of about 12.5 to 75 μm (0.5 to 3 mils), and preferably about 25 to 50 μm (1 to 2 mils).

The conductive traces are typically formed using a conductive material 56 such as carbon (e.g., graphite), a conductive polymer, a metal or alloy (e.g., gold or gold alloy), or a metallic compound (e.g., ruthenium dioxide or titanium dioxide). The formation of films of carbon, conductive polymer, metal, alloy, or metallic compound are well-known and include, for example, chemical vapor deposition (CVD), physical vapor deposition, sputtering, reactive sputtering, printing, coating, and painting. The conductive material 56 which fills the channels 54 is often formed using a precursor material, such as a conductive ink or paste. In these embodiments, the conductive material 56 is deposited on the substrate 50 using methods such as coating, painting, or applying the material using a spreading instrument, such as a coating blade. Excess conductive material between the channels 54 is then removed by, for example, running a blade along the substrate surface.

In one embodiment, the conductive material 56 is a part of a precursor material, such as a conductive ink, obtainable, for example, from Ercon, Inc. (Wareham, Mass.), Metech, Inc. (Elverson, Pa.), E.I. du Pont de Nemours and Co. (Wilmington, Del.), Emca-Remex Products (Montgomeryville, Pa.), or MCA Services (Melbourn, Great Britain). The conductive ink is typically applied as a semi-liquid or paste which contains particles of the carbon, metal, alloy, or metallic compound and a solvent or dispersant. After application of the conductive ink on the substrate 50 (e.g., in the channels 54), the solvent or dispersant evaporates to leave behind a solid mass of conductive material 56.

In addition to the particles of carbon, metal, alloy, or metallic compound, the conductive ink may also contain a binder. The binder may optionally be cured to further bind the conductive material 56 within the channel 54 and/or on the substrate 50. Curing the binder increases the conductivity of the conductive material 56. However, this is typically not necessary as the currents carried by the conductive material 56 within the conductive traces 52 are often rela-

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tively low (usually less than 1 μA and often less than 100 nA). Typical binders include, for example, polyurethane resins, cellulose derivatives, elastomers, and highly fluorinated polymers. Examples of elastomers include silicones, polymeric dienes, and acrylonitrile-butadiene-styrene (ABS) resins. One example of a fluorinated polymer binder is Teflon® (DuPont, Wilmington, Del.). These binders are cured using, for example, heat or light, including ultraviolet (UV) light. The appropriate curing method typically depends on the particular binder which is used.

Often, when a liquid or semiliquid precursor of the conductive material 56 (e.g., a conductive ink) is deposited in the channel 54, the precursor fills the channel 54. However, when the solvent or dispersant evaporates, the conductive material 56 which remains may lose volume such that the conductive material 56 may or may not continue to fill the channel 54. Preferred conductive materials 56 do not pull away from the substrate 50 as they lose volume, but rather decrease in height within the channel 54. These conductive materials 56 typically adhere well to the substrate 50 and therefore do not pull away from the substrate 50 during evaporation of the solvent or dispersant. Other suitable conductive materials 56 either adhere to at least a portion of the substrate 50 and/or contain another additive, such as a binder, which adheres the conductive material 56 to the substrate 50. Preferably, the conductive material 56 in the channels 54 is non-leachable, and more preferably immobilized on the substrate 50. In some embodiments, the conductive material 56 may be formed by multiple applications of a liquid or semiliquid precursor interspersed with removal of the solvent or dispersant.

In another embodiment, the channels 54 are formed using a laser. The laser carbonizes the polymer or plastic material. The carbon formed in this process is used as the conductive material 56. Additional conductive material 56, such as a conductive carbon ink, may be used to supplement the carbon formed by the laser.

In a further embodiment, the conductive traces 52 are formed by pad printing techniques. For example, a film of conductive material is formed either as a continuous film or as a coating layer deposited on a carrier film. This film of conductive material is brought between a print head and the substrate 50. A pattern on the surface of the substrate 50 is made using the print head according to a desired pattern of conductive traces 52. The conductive material is transferred by pressure and/or heat from the film of conductive material to the substrate 50. This technique often produces channels (e.g., depressions caused by the print head) in the substrate 50. Alternatively, the conductive material is deposited on the surface of the substrate 50 without forming substantial depressions.

In other embodiments, the conductive traces 52 are formed by non-impact printing techniques. Such techniques include electrophotography and magnetography. In these processes, an image of the conductive traces 52 is electrically or magnetically formed on a drum. A laser or LED may be used to electrically form an image. A magnetic recording head may be used to magnetically form an image. A toner material (e.g., a conductive material, such as a conductive ink) is then attracted to portions of the drum according to the image. The toner material is then applied to the substrate by contact between the drum and the substrate. For example, the substrate may be rolled over the drum. The toner material may then be dried and/or a binder in the toner material may be cured to adhere the toner material to the substrate.

Another non-impact printing technique includes ejecting droplets of conductive material onto the substrate in a

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desired pattern. Examples of this technique include ink jet printing and piezo jet printing. An image is sent to the printer which then ejects the conductive material (e.g., a conductive ink) according to the pattern. The printer may provide a continuous stream of conductive material or the printer may eject the conductive material in discrete amounts at the desired points.

Yet another non-impact printing embodiment of forming the conductive traces includes an ionographic process. In the this process, a curable, liquid precursor, such as a photopolymerizable acrylic resin (e.g., Solimer 7501 from Cubital, Bad Kreuznach, Germany) is deposited over a surface of a substrate 50. A photomask having a positive or negative image of the conductive traces 52 is then used to cure the liquid precursor. Light (e.g., visible or ultraviolet light) is directed through the photomask to cure the liquid precursor and form a solid layer over the substrate according to the image on the photomask. Uncured liquid precursor is removed leaving behind channels 54 in the solid layer. These channels 54 can then be filled with conductive material 56 to form conductive traces 52.

Conductive traces 52 (and channels 54, if used) can be formed with relatively narrow widths, for example, in the range of 25 to 250 μm, and including widths of, for example, 250 μm, 150 μm, 100 μm, 75 μm, 50 μm, 25 μm or less by the methods described above. In embodiments with two or more conductive traces 52 on the same side of the substrate 50, the conductive traces 52 are separated by distances sufficient to prevent conduction between the conductive traces 52. The edge-to-edge distance between the conductive traces is preferably in the range of 25 to 250 μm and may be, for example, 150 μm, 100 μm, 75 μm, 50 μm, or less. The density of the conductive traces 52 on the substrate 50 is preferably in the range of about 150 to 700 μm/trace and may be as small as 667 μm/trace or less, 333 μm/trace or less, or even 167 μm/trace or less.

The working electrode 58 and the counter electrode 60 (if a separate reference electrode is used) are often made using a conductive material 56, such as carbon. Suitable carbon conductive inks are available from Ercon, Inc. (Wareham, Mass.), Metech, Inc. (Elverson, Pa.), E.I. du Pont de Nemours and Co. (Wilmington, Del.), Emca-Remex Products (Montgomeryville, Pa.), or MCA Services (Melbourn, Great Britain). Typically, the working surface 51 of the working electrode 58 is at least a portion of the conductive trace 52 that is in contact with the analyte-containing fluid (e.g., implanted in the patient).

The reference electrode 62 and/or counter/reference electrode are typically formed using conductive material 56 that is a suitable reference material, for example silver/silver chloride or a non-leachable redox couple bound to a conductive material, for example, a carbon-bound redox couple. Suitable silver/silver chloride conductive inks are available from Ercon, Inc. (Wareham, Mass.), Metech, Inc. (Elverson, Pa.), E.I. du Pont de Nemours and Co. (Wilmington, Del.), Emca-Remex Products (Montgomeryville, Pa.), or MCA Services (Melbourn, Great Britain). Silver/silver chloride electrodes illustrate a type of reference electrode that involves the reaction of a metal electrode with a constituent of the sample or body fluid, in this case, Cl$^-$.

Suitable redox couples for binding to the conductive material of the reference electrode include, for example, redox polymers (e.g., polymers having multiple redox centers.) It is preferred that the reference electrode surface be non-corroding so that an erroneous potential is not measured. Preferred conductive materials include less corrosive metals, such as gold and palladium. Most preferred

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are non-corrosive materials including non-metallic conductors, such as carbon and conducting polymers. A redox polymer can be adsorbed on or covalently bound to the conductive material of the reference electrode, such as a carbon surface of a conductive trace **52**. Non-polymeric redox couples can be similarly bound to carbon or gold surfaces.

A variety of methods may be used to immobilize a redox polymer on an electrode surface. One method is adsorptive immobilization. This method is particularly useful for redox polymers with relatively high molecular weights. The molecular weight of a polymer may be increased, for example, by cross-linking.

Another method for immobilizing the redox polymer includes the functionalization of the electrode surface and then the chemical bonding, often covalently, of the redox polymer to the functional groups on the electrode surface. One example of this type of immobilization begins with a poly(4-vinylpyridine). The polymer's pyridine rings are, in part, complexed with a reducible/oxidizable species, such as $[Os(bpy)_2Cl]^{+/2+}$ where bpy is 2,2'-bipyridine. Part of the pyridine rings are quaternized by reaction with 2-bromoethylamine. The polymer is then crosslinked, for example, using a diepoxide, such as polyethylene glycol diglycidyl ether.

Carbon surfaces can be modified for attachment of a redox species or polymer, for example, by electroreduction of a diazonium salt. As an illustration, reduction of a diazonium salt formed upon diazotization of p-aminobenzoic acid modifies a carbon surface with phe-nylcarboxylic acid functional groups. These functional groups can then be activated by a carbodiimide, such as 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydro-chloride. The activated functional groups are then bound with a amine-functionalized redox couple, such as the quat-ernized osmium-containing redox polymer described above or 2-aminoethylferrocene, to form the redox couple.

Similarly, gold can be functionalized by an amine, such as cystamine,. A redox couple such as $[Os(bpy)_2(pyridine-4-carboxylate)Cl]^{0/+}$ is activated by 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydrochloride to form a reactive O-acylisourea which reacts with the gold-bound amine to form an amide.

In one embodiment, in addition to using the conductive traces **52** as electrodes or probe leads, two or more of the conductive traces **52** on the substrate **50** are used to give the patient a mild electrical shock when, for example, the analyte level exceeds a threshold level. This shock may act as a warning or alarm to the patient to initiate some action to restore the appropriate level of the analyte.

The mild electrical shock is produced by applying a potential between any two conductive traces **52** that are not otherwise connected by a conductive path. For example, two of the electrodes **58, 60, 62** or one electrode **58, 60, 62** and the temperature probe **66** may be used to provide the mild shock. Preferably, the working electrode **58** and the refer-ence electrode **62** are not used for this purpose as this may cause some damage to the chemical components on or proximate to the particular electrode (e.g., the sensing layer on the working electrode or the redox couple on the refer-ence electrode).

The current used to produce the mild shock is typically 0.1 to 1 mA. Higher or lower currents may be used, although care should be taken to avoid harm to the patient. The potential between the conductive traces is typically 1 to 10 volts. However, higher or lower voltages may be used depending, for example, on the resistance of the conductive

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traces **52**, the distance between the conductive traces **52** and the desired amount of current. When the mild shock is delivered, potentials at the working electrode **58** and across the temperature probe **66** may be removed to prevent harm to those components caused by unwanted conduction between the working electrode **58** (and/or temperature probe **66**, if used) and the conductive traces **52** which provide the mild shock.

Contact Pads

Typically, each of the conductive traces **52** includes a contact pad **49**. The contact pad **49** may simply be a portion of the conductive trace **52** that is indistinguishable from the rest of the trace **52** except that the contact pad **49** is brought into contact with the conductive contacts of a control unit (e.g., the sensor control unit **44** of FIG. **1**). More commonly, however, the contact pad **49** is a region of the conductive trace **52** that has a larger width than other regions of the trace **52** to facilitate a connection with the contacts on the control unit. By making the contact pads **49** relatively large as compared with the width of the conductive traces **52**, the need for precise registration between the contact pads **49** and the contacts on the control unit is less critical than with small contact pads.

The contact pads **49** are typically made using the same material as the conductive material **56** of the conductive traces **52**. However, this is not necessary. Although metal, alloys, and metallic compounds may be used to form the contact pads **49**, in some embodiments, it is desirable to make the contact pads **49** from a carbon or other non-metallic material, such as a conducting polymer. In contrast to metal or alloy contact pads, carbon and other non-metallic contact pads are not easily corroded if the contact pads **49** are in a wet, moist, or humid environment. Metals and alloys may corrode under these conditions, particularly if the contact pads **49** and contacts of the control unit are made using different metals or alloys. However, carbon and non-metallic contact pads **49** do not significantly corrode, even if the contacts of the control device are metal or alloy.

One embodiment of the invention includes a sensor **42** having contact pads **49** and a control unit **44** having con-ductive contacts (not shown). During operation of the sensor **42**, the contact pads **49** and conductive contacts are in contact with each other. In this embodiment, either the contact pads **49** or the conductive contacts are made using a non-corroding, conductive material. Such materials include, for example, carbon and conducting polymers. Preferred non-corroding materials include graphite and vitreous car-bon. The opposing contact pad or conductive contact is made using carbon, a conducting polymer, a metal, such as gold, palladium, or platinum group metal, or a metallic compound, such as ruthenium dioxide. This configuration of contact pads and conductive contacts typically reduces cor-rosion. Preferably, when the sensor is placed in a 3 mM, and more preferably, in a 100 mM, NaCl solution, the signal arising due to the corrosion of the contact pads and/or conductive contacts is less than 3% of the signal generated by the sensor when exposed to concentration of analyte in the normal physiological range. For at least some subcuta-neous glucose sensors, the current generated by analyte in a normal physiological range ranges from 3 to 500 nA.

Each of the electrodes **58, 60, 62**, as well as the two probe leads **68, 70** of the temperature probe **66** (described below), are connected to contact pads **49** as shown in FIGS. **10** and **11**. In one embodiment (not shown), the contact pads **49** are on the same side of the substrate **50** as the respective electrodes or temperature probe leads to which the contact pads **49** are attached.

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In other embodiments, the conductive traces **52** on at least one side are connected through vias in the substrate to contact pads **49a** on the opposite surface of the substrate **50**, as shown in FIGS. **10** and **11**. An advantage of this configuration is that contact between the contacts on the control unit and each of the electrodes **58, 60, 62** and the probe leads **68,70** of the temperature probe **66** can be made from a single side of the substrate **50**.

In yet other embodiments (not shown), vias through the substrate are used to provide contact pads on both sides of the substrate **50** for each conductive trace **52**. The vias connecting the conductive traces **52** with the contact pads **49a** can be formed by making holes through the substrate **50** at the appropriate points and then filling the holes with conductive material **56**.

Exemplary Electrode Configurations

A number of exemplary electrode configurations are described below, however, it will be understood that other configurations may also be used. In one embodiment, illustrated in FIG. 3A, the sensor **42** includes two working electrodes **58a, 58b** and one counter electrode **60**, which also functions as a reference electrode. In another embodiment, the sensor includes one working electrode **58a,** one counter electrode **60**, and one reference electrode **62**, as shown in FIG. 3B. Each of these embodiments is illustrated with all of the electrodes formed on the same side of the substrate **50**.

Alternatively, one or more of the electrodes may be formed on an opposing side of the substrate **50**. This may be convenient if the electrodes are formed using two different types of conductive material **56** (e.g., carbon and silver/silver chloride). Then, at least in some embodiments, only one type of conductive material **56** needs to be applied to each side of the substrate **50**, thereby reducing the number of steps in the manufacturing process and/or easing the registration constraints in the process. For example, if the working electrode **58** is formed using a carbon-based conductive material **56** and the reference or counter/reference electrode is formed using a silver/silver chloride conductive material **56**, then the working electrode and reference or counter/reference electrode may be formed on opposing sides of the substrate **50** for case of manufacture.

In another embodiment, two working electrodes **58** and one counter electrode **60** are formed on one side of the substrate **50** and one reference electrode **62** and a temperature probe **66** are formed on an opposing side of the substrate **50**, as illustrated in FIG. **6**. The opposing sides of the tip of this embodiment of the sensor **42** are illustrated in FIGS. **7** and **8**.

Sensing Layer

Some analytes, such as oxygen, can be directly electrooxidized or electroreduced on the working electrode **58**. Other analytes, such as glucose and lactate, require the presence of at least one electron transfer agent and/or at least one catalyst to facilitate the electrooxidation or electroreduction of the analyte. Catalysts may also be used for those analyte, such as oxygen, that can be directly electrooxidized or electroreduced on the working electrode **58**. For these analytes, each working electrode **58** has a sensing layer **64** formed proximate to or on a working surface of the working electrode **58**. Typically, the sensing layer **64** is formed near or on only a small portion of the working electrode **58**, often near a tip of the sensor **42**. This limits the amount of material needed to form the sensor **42** and places the sensing layer **64** in the best position for contact with the analyte-containing fluid (e.g., a body fluid, sample fluid, or carrier fluid).

The sensing layer **64** includes one or more components designed to facilitate the electrolysis of the analyte. The sensing layer **64** may include, for example, a catalyst to catalyze a reaction of the analyte and produce a response at the working electrode **58**, an electron transfer agent to indirectly or directly transfer electrons between the analyte and the working electrode **58**, or both.

The sensing layer **64** may be formed as a solid composition of the desired components (e.g., an electron transfer agent and/or a catalyst). These components are preferably non-leachable from the sensor **42** and more preferably are immobilized on the sensor **42**. For example, the components may be immobilized on a working electrode **58**. Alternatively, the components of the sensing layer **64** may be immobilized within or between one or more membranes or films disposed over the working electrode **58** or the components may be immobilized in a polymeric or sol-gel matrix. Examples of immobilized sensing layers are described in U.S. Pat. Nos. 5,262,035, 5,264,104, 5,264,105, 5,320,725, 5,593,852, and 5,665,222, U.S. patent application Ser. No. 08/540,789, and PCT Patent Application No. US98/02403 entitled "Soybean Peroxidase Electrochemical Sensor", filed on Feb. 11, 1998, incorporated herein by reference.

In some embodiments, one or more of the components of the sensing layer **64** may be solvated, dispersed, or suspended in a fluid within the sensing layer **64**, instead of forming a solid composition. The fluid may be provided with the sensor **42** or may be absorbed by the sensor **42** from the analyte-containing fluid. Preferably, the components which are solvated, dispersed, or suspended in this type of sensing layer **64** are non-leachable from the sensing layer. Non-leachability may be accomplished, for example, by providing barriers (e.g., the electrode, substrate, membranes, and/or films) around the sensing layer which prevent the leaching of the components of the sensing layer **64**. One example of such a barrier is a microporous membrane or film which allows diffusion of the analyte into the sensing layer **64** to make contact with the components of the sensing layer **64**, but reduces or eliminates the diffusion of the sensing layer components (e.g., a electron transfer agent and/or a catalyst) out of the sensing layer **64**.

A variety of different sensing layer configurations can be used. In one embodiment, the sensing layer **64** is deposited on the conductive material **56** of a working electrode **58a,** as illustrated in FIGS. **3A** and **3B**. The sensing layer **64** may extend beyond the conductive material **56** of the working electrode **58a**. In some cases, the sensing layer **64** may also extend over the counter electrode **60** or reference electrode **62** without degrading the performance of the glucose sensor. For those sensors **42** which utilize channels **54** within which the conductive material **56** is deposited, a portion of the sensing layer **64** may be formed within the channel **54** if the conductive material **56** does not fill the channel **54**.

A sensing layer **64** in direct contact with the working electrode **58a** may contain an electron transfer agent to transfer electrons directly or indirectly between the analyte and the working electrode, as well as a catalyst to facilitate a reaction of the analyte. For example, a glucose, lactate, or oxygen electrode may be formed having a sensing layer which contains a catalyst, such as glucose oxidase, lactate oxidase, or laccase, respectively, and an electron transfer agent that facilitates the electrooxidation of the glucose, lactate, or oxygen, respectively.

In another embodiment, the sensing layer **64** is not deposited directly on the working electrode **58a**. Instead, the sensing layer **64** is spaced apart from the working electrode **58a,** as illustrated in FIG. **4A**, and separated from the working electrode **58a** by a separation layer **61**. The sepa-

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ration layer **61** typically includes one or more membranes or films. In addition to separating the working electrode **58**a from the sensing layer **64**, the separation layer **61** may also act as a mass transport limiting layer or an interferent eliminating layer, as described below.

Typically, a sensing layer **64**, which is not in direct contact with the working electrode **58**a, includes a catalyst that facilitates a reaction of the analyte. However, this sensing layer **64** typically does not include an electron transfer agent that transfers electrons directly from the working electrode **58**a to the analyte, as the sensing layer **64** is spaced apart from the working electrode **58**a. One example of this type of sensor is a glucose or lactate sensor which includes an enzyme (e.g., glucose oxidase or lactate oxidase, respectively) in the sensing layer **64**. The glucose or lactate reacts with a second compound (e.g., oxygen) in the presence of the enzyme. The second compound is then electrooxidized or electroreduced at the electrode. Changes in the signal at the electrode indicate changes in the level of the second compound in the fluid and are proportional to changes in glucose or lactate level and, thus, correlate to the analyte level.

In another embodiment, two sensing layers **63**, **64** are used, as shown in FIG. **4B**. Each of the two sensing layers **63**, **64** may be independently formed on the working electrode **58**a or in proximity to the working electrode **58**a. One sensing layer **64** is typically, although not necessarily, spaced apart from the working electrode **58**a. For example, this sensing layer **64** may include a catalyst which catalyzes a reaction of the analyte to form a product compound. The product compound is then electrolyzed in the second sensing layer **63** which may include an electron transfer agent to transfer electrons between the working electrode **58**a and the product compound and/or a second catalyst to catalyze a reaction of the product compound to generate a signal at the working electrode **58**a.

For example, a glucose or lactate sensor may include a first sensing layer **64** which is spaced apart from the working electrode and contains an enzyme, for example, glucose oxidase or lactate oxidase. The reaction of glucose or lactate in the presence of the appropriate enzyme forms hydrogen peroxide. A second sensing layer **63** is provided directly on the working electrode **58**a and contains a peroxidase enzyme and an electron transfer agent to generate a signal at the electrode in response to the hydrogen peroxide. The level of hydrogen peroxide indicated by the sensor then correlates to the level of glucose or lactate. Another sensor which operates similarly can be made using a single sensing layer with both the glucose or lactate oxidase and the peroxidase being deposited in the single sensing layer. Examples of such sensors are described in U.S. Pat. No. 5,593,852, U.S. patent application Ser. No. 08/540,789, and PCT Patent Application No. US98/02403 entitled "Soybean Peroxidase Electrochemical Sensor", filed on Feb. 11, 1998, incorporated herein by reference.

In some embodiments, one or more of the working electrodes **58**b do not have a corresponding sensing layer **64**, as shown in FIGS. **3A** and **4A**, or have a sensing layer (not shown) which does not contain one or more components (e.g., an electron transfer agent or catalyst) needed to electrolyze the analyte. The signal generated at this working electrode **58**b typically arises from interferents and other sources, such as ions, in the fluid, and not in response to the analyte (because the analyte is not electrooxidized or electroreduced). Thus, the signal at this working electrode **58**b corresponds to a background signal. The background signal can be removed from the analyte signal obtained from

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other working electrodes **58**a that are associated with fully-functional sensing layers **64** by, for example, subtracting the signal at working electrode **58**b from the signal at working electrode **58**a.

Sensors having multiple working electrodes **58**a may also be used to obtain more precise results by averaging the signals or measurements generated at these working electrodes **58**a. In addition, multiple readings at a single working electrode **58**a or at multiple working electrodes may be averaged to obtain more precise data.

Electron Transfer Agent

In many embodiments, the sensing layer **64** contains one or more electron transfer agents in contact with the conductive material **56** of the working electrode **58**, as shown in FIGS. **3A** and **3B**. In some embodiments of the invention, there is little or no leaching of the electron transfer agent away from the working electrode **58** during the period in which the sensor **42** is implanted in the patient. A diffusing or leachable (i.e., releasable) electron transfer agent often diffuses into the analyte-containing fluid, thereby reducing the effectiveness of the electrode by reducing the sensitivity of the sensor over time. In addition, a diffusing or leaching electron transfer agent in an implantable sensor **42** may also cause damage to the patient. In these embodiments, preferably, at least 90%, more preferably, at least 95%, and, most preferably, at least 99%, of the electron transfer agent remains disposed on the sensor after immersion in the analyte-containing fluid for 24 hours, and, more preferably, for 72 hours. In particular, for an implantable sensor, preferably, at least 90%, more preferably, at least 95%, and most preferably, at least 99%, of the electron transfer agent remains disposed on the sensor after immersion in the body fluid at 37° C. for 24 hours, and, more preferably, for 72 hours.

In some embodiments of the invention, to prevent leaching, the electron transfer agents are bound or otherwise immobilized on the working electrode **58** or between or within one or more membranes or films disposed over the working electrode **58**. The electron transfer agent may be immobilized on the working electrode **58** using, for example, a polymeric or sol-gel immobilization technique. Alternatively, the electron transfer agent may be chemically (e.g., ionically, covalently, or coordinatively) bound to the working electrode **58**, either directly or indirectly through another molecule, such as a polymer, that is in turn bound to the working electrode **58**.

Application of the sensing layer **64** on a working electrode **58**a is one method for creating a working surface for the working electrode **58**a, as shown in FIGS. **3A** and **3B**. The electron transfer agent mediates the transfer of electrons to electrooxidize or electroreduce an analyte and thereby permits a current flow between the working electrode **58** and the counter electrode **60** via the analyte. The mediation of the electron transfer agent facilitates the electrochemical analysis of analytes which are not suited for direct electrochemical reaction on an electrode.

In general, the preferred electron transfer agents are electroreducible and electrooxidizable ions or molecules having redox potentials that are a few hundred millivolts above or below the redox potential of the standard calomel electrode (SCE). Preferably, the electron transfer agents are not more reducing than about −150 mV and not more oxidizing than about +400 mV versus SCE.

The electron transfer agent may be organic, organometallic, or inorganic. Examples of organic redox species are quinones and species that in their oxidized state have quinoid structures, such as Nile blue and indophenol.

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Some quinones and partially oxidized quinhydrones react with functional groups of proteins such as the thiol groups of cysteine, the amine groups of lysine and arginine, and the phenolic groups of tyrosine which may render those redox species unsuitable for some of the sensors of the present invention because of the presence of the interfering proteins in an analyte-containing fluid. Usually substituted quinones and molecules with quinoid structure are less reactive with proteins and are preferred. A preferred tetrasubstituted quinone usually has carbon atoms in positions 1, 2, 3, and 4.

In general, electron transfer agents suitable for use in the invention have structures or charges which prevent or substantially reduce the diffusional loss of the electron transfer agent during the period of time that the sample is being analyzed. The preferred electron transfer agents include a redox species bound to a polymer which can in turn be immobilized on the working electrode. The bond between the redox species and the polymer may be covalent, coordinative, or ionic. Useful electron transfer agents and methods for producing them are described in U.S. Pat. Nos. 5,264,104; 5,356,786; 5,262,035; and 5,320,725, incorporated herein by reference. Although any organic or organometallic redox species can be bound to a polymer and used as an electron transfer agent, the preferred redox species is a transition metal compound or complex. The preferred transition metal compounds or complexes include osmium, ruthenium, iron, and cobalt compounds or complexes. The most preferred are osmium compounds and complexes. It will be recognized that many of the redox species described below may also be used, typically without a polymeric component, as electron transfer agents in a carrier fluid or in a sensing layer of a sensor where leaching of the electron transfer agent is acceptable.

One type of non-releasable polymeric electron transfer agent contains a redox species covalently bound in a polymeric composition. An example of this type of mediator is poly(vinylferrocene).

Another type of non-releasable electron transfer agent contains an ionically-bound redox species. Typically, this type of mediator includes a charged polymer coupled to an oppositely charged redox species. Examples of this type of mediator include a negatively charged polymer such as Nafion® (DuPont) coupled to a positively charged redox species such as an osmium or ruthenium polypyridyl cation. Another example of an ionically-bound mediator is a positively charged polymer such as quaternized poly(4-vinyl pyridine) or poly(1-vinyl imidazole) coupled to a negatively charged redox species such as ferricyanide or ferrocyanide. The preferred ionically-bound redox species is a highly charged redox species bound within an oppositely charged redox polymer.

In another embodiment of the invention, suitable non-releasable electron transfer agents include a redox species coordinatively bound to a polymer. For example, the mediator may be formed by coordination of an osmium or cobalt 2,2'-bipyridyl complex to poly(1-vinyl imidazole) or poly(4-vinyl pyridine).

The preferred electron transfer agents are osmium transition metal complexes with one or more ligands, each ligand having a nitrogen-containing heterocycle such as 2,2'-bipyridine, 1,10-phenanthroline, or derivatives thereof. Furthermore, the preferred electron transfer agents also have one or more ligands covalently bound in a polymer, each ligand having at least one nitrogen-containing heterocycle, such as pyridine, imidazole, or derivatives thereof. These preferred electron transfer agents exchange electrons rapidly between each other and the working electrodes **58** so that the complex can be rapidly oxidized and reduced.

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One example of a particularly useful electron transfer agent includes (a) a polymer or copolymer having pyridine or imidazole functional groups and (b) osmium cations complexed with two ligands, each ligand containing 2,2'-bipyridine, 1,10-phenanthroline, or derivatives thereof, the two ligands not necessarily being the same. Preferred derivatives of 2,2'-bipyridine for complexation with the osmium cation are 4,4'-dimethyl-2,2'-bipyridine and mono-, di-, and polyalkoxy-2,2'-bipyridines, such as 4,4'-dimethoxy-2,2'-bipyridine. Preferred derivatives of 1,10-phenanthroline for complexation with the osmium cation are 4,7-dimethyl-1, 10-phenanthroline and mono, di-, and polyalkoxy-1,10-phenanthrolines, such as 4,7-dimethoxy-1,10-phenanthroline. Preferred polymers for complexation with the osmium cation include polymers and copolymers of poly(1-vinyl imidazole) (referred to as "PVI") and poly(4-vinyl pyridine) (referred to as "PVP"). Suitable copolymer substituents of poly(1-vinyl imidazole) include acrylonitrile, acrylamide, and substituted or quaternized N-vinyl imidazole. Most preferred are electron transfer agents with osmium complexed to a polymer or copolymer of poly(1-vinyl imidazole).

The preferred electron transfer agents have a redox potential ranging from −100 mV to about +150 mV versus the standard calomel electrode (SCE). Preferably, the potential of the electron transfer agent ranges from −100 mV to +150 mV and more preferably, the potential ranges from −50 mV to +50 mV. The most preferred electron transfer agents have osmium redox centers and a redox potential ranging from +50 mV to −150 mV versus SCE.

Catalyst

The sensing layer **64** may also include a catalyst which is capable of catalyzing a reaction of the analyte. The catalyst may also, in some embodiments, act as an electron transfer agent. One example of a suitable catalyst is an enzyme which catalyzes a reaction of the analyte. For example, a catalyst, such as a glucose oxidase, glucose dehydrogenase (e.g., pyrroloquinoline quinone glucose dehydrogenase (PQQ)), or oligosaccharide dehydrogenase, may be used when the analyte is glucose. A lactate oxidase or lactate dehydrogenase may be used when the analyte is lactate. Laccase may be used when the analyte is oxygen or when oxygen is generated or consumed in response to a reaction of the analyte.

Preferably, the catalyst is non-leachably disposed on the sensor, whether the catalyst is part of a solid sensing layer in the sensor or solvated in a fluid within the sensing layer. More preferably, the catalyst is immobilized within the sensor (e.g., on the electrode and/or within or between a membrane or film) to prevent unwanted leaching of the catalyst away from the working electrode **58** and into the patient. This may be accomplished, for example, by attaching the catalyst to a polymer, cross linking the catalyst with another electron transfer agent (which, as described above, can be polymeric), and/or providing one or more barrier membranes or films with pore sizes smaller than the catalyst.

As described above, a second catalyst may also be used. This second catalyst is often used to catalyze a reaction of a product compound resulting from the catalyzed reaction of the analyte. The second catalyst typically operates with an electron transfer agent to electrolyze the product compound to generate a signal at the working electrode. Alternatively, the second catalyst may be provided in an interferent-eliminating layer to catalyze reactions that remove interferents, as described below.

One embodiment of the invention is an electrochemical sensor in which the catalyst is mixed or dispersed in the

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conductive material **56** which forms the conductive trace **52** of a working electrode **58**. This may be accomplished, for example, by mixing a catalyst, such as an enzyme, in a carbon ink and applying the mixture into a channel **54** on the surface of the substrate **50**. Preferably, the catalyst is immobilized in the channel **53** so that it can not leach away from the working electrode **58**. This may be accomplished, for example, by curing a binder in the carbon ink using a curing technique appropriate to the binder. Curing techniques include, for example, evaporation of a solvent or dispersant, exposure to ultraviolet light, or exposure to heat. Typically, the mixture is applied under conditions that do not substantially degrade the catalyst. For example, the catalyst may be an enzyme that is heat-sensitive. The enzyme and conductive material mixture should be applied and cured, preferably, without sustained periods of heating. The mixture may be cured using evaporation or UV curing techniques or by the exposure to heat that is sufficiently short that the catalyst is not substantially degraded.

Another consideration for in vivo analyte sensors is the thermostability of the catalyst. Many enzymes have only limited stability at biological temperatures. Thus, it may be necessary to use large amounts of the catalyst and/or use a catalyst that is thermostable at the necessary temperature (e.g., 37° C. or higher for normal body temperature). A thermostable catalyst may be defined as a catalyst which loses less than 5% of its activity when held at 37° C. for at least one hour, preferably, at least one day, and more preferably at least three days. One example of a thermostable catalyst is soybean peroxidase. This particular thermostable catalyst may be used in a glucose or lactate sensor when combined either in the same or separate sensing layers with glucose or lactate oxidase or dehydrogenase. A further description of thermostable catalysts and their use in electrochemical inventions is found in U.S. Pat. No. 5,665,222 U.S. patent application Ser. No. 08/540,789, and PCT Application No. US98/02403 entitled "Soybean Peroxidase Electrochemical Sensor", filed on Feb. 11, 1998.

Electrolysis of the Analyte

To electrolyze the analyte, a potential (versus a reference potential) is applied across the working and counter electrodes **58**, **60**. The minimum magnitude of the applied potential is often dependent on the particular electron transfer agent, analyte (if the analyte is directly electrolyzed at the electrode), or second compound (if a second compound, such as oxygen or hydrogen peroxide, whose level is dependent on the analyte level, is directly electrolyzed at the electrode). The applied potential usually equals or is more oxidizing or reducing, depending on the desired electrochemical reaction, than the redox potential of the electron transfer agent, analyte, or second compound, whichever is directly electrolyzed at the electrode. The potential at the working electrode is typically large enough to drive the electrochemical reaction to or near completion.

The magnitude of the potential may optionally be limited to prevent significant (as determined by the current generated in response to the analyte) electrochemical reaction of interferents, such as urate, ascorbate, and acetaminophen. The limitation of the potential may be obviated if these interferents have been removed in another way, such as by providing an interferent-limiting barrier, as described below, or by including a working electrode **58b** (see FIG. **3A**) from which a background signal may be obtained.

When a potential is applied between the working electrode **58** and the counter electrode **60**, an electrical current will flow. The current is a result of the electrolysis of the analyte or a second compound whose level is affected by the

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analyte. In one embodiment, the electrochemical reaction occurs via an electron transfer agent and the optional catalyst. Many analytes B are oxidized (or reduced) to products C by an electron transfer agent species A in the presence of an appropriate catalyst (e.g., an enzyme). The electron transfer agent A is then oxidized (or reduced) at the electrode. Electrons are collected by (or removed from) the electrode and the resulting current is measured. This process is illustrated by reaction equations (1) and (2) (similar equations may be written for the reduction of the analyte B by a redox mediator A in the presence of a catalyst):

$$nA(ox) + B \xrightarrow{\text{catalyst}} nA(red) + C \tag{1}$$

$$nA(red) \xrightarrow{\text{electrode}} nA(ox) + ne^- \tag{2}$$
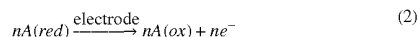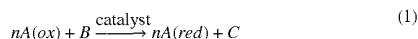
As an example, an electrochemical sensor may be based on the reaction of a glucose molecule with two nonleachable ferricyanide anions in the presence of glucose oxidase to produce two non-leachable ferrocyanide anions, two hydrogen ions, and gluconolactone. The amount of glucose present is assayed by electrooxidizing the nonleachable ferrocyanide anions to non-leachable ferricyanide anions and measuring the current.

In another embodiment, a second compound whose level is affected by the analyte is electrolyzed at the working electrode. In some cases, the analyte D and the second compound, in this case, a reactant compound E, such as oxygen, react in the presence of the catalyst, as shown in reaction equation (3).

$$D + E \xrightarrow{\text{catalyst}} F + G \tag{3}$$

The reactant compound E is then directly oxidized (or reduced) at the working electrode, as shown in reaction equation (4)

$$nE(red) \xrightarrow{\text{electrode}} nE(ox) + ne^- \tag{4}$$
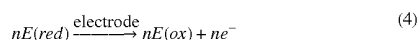
Alternatively, the reactant compound E is indirectly oxidized (or reduced) using an electron transfer agent H (optionally in the presence of a catalyst), that is subsequently reduced or oxidized at the electrode, as shown in reaction equations (5) and (6).

$$nH(ox) + E \rightarrow nH(red) + I \tag{5}$$

$$nH(red) \xrightarrow{\text{electrode}} nH(ox) + ne^- \tag{6}$$
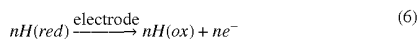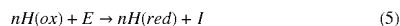
In either case, changes in the concentration of the reactant compound, as indicated by the signal at the working electrode, correspond inversely to changes in the analyte (i.e., as the level of analyte increase then the level of reactant compound and the signal at the electrode decreases.)

In other embodiments, the relevant second compound is a product compound F, as shown in reaction equation (3). The product compound F is formed by the catalyzed reaction of analyte D and then be directly electrolyzed at the electrode or indirectly electrolyzed using an electron transfer agent and, optionally, a catalyst. In these embodiments, the signal arising from the direct or indirect electrolysis of the product compound F at the working electrode corresponds directly to the level of the analyte (unless there are other sources of the

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product compound). As the level of analyte increases, the level of the product compound and signal at the working electrode increases.

Those skilled in the art will recognize that there are many different reactions that will achieve the same result; namely the electrolysis of an analyte or a compound whose level depends on the level of the analyte. Reaction equations (1) through (6) illustrate non-limiting examples of such reactions.

Temperature Probe

A variety of optional items may be included in the sensor. One optional item is a temperature probe 66 (FIGS. 8 and 11). The temperature probe 66 may be made using a variety of known designs and materials. One exemplary temperature probe 66 is formed using two probe leads 68, 70 connected to each other through a temperature-dependent element 72 that is formed using a material with a temperature-dependent characteristic. An example of a suitable temperature-dependent characteristic is the resistance of the temperature-dependent element 72.

The two probe leads 68, 70 are typically formed using a metal, an alloy, a semimetal, such as graphite, a degenerate or highly doped semiconductor, or a small-band gap semiconductor. Examples of suitable materials include gold, silver, ruthenium oxide, titanium nitride, titanium dioxide, indium doped tin oxide, tin doped indium oxide, or graphite. The temperature-dependent element 72 is typically made using a fine trace (e.g., a conductive trace that has a smaller cross-section than that of the probe leads 68, 70) of the same conductive material as the probe leads, or another material such as a carbon ink, a carbon fiber, or platinum, which has a temperature-dependent characteristic, such as resistance, that provides a temperature-dependent signal when a voltage source is attached to the two probe leads 68, 70 of the temperature probe 66. The temperature-dependent characteristic of the temperature-dependent element 72 may either increase or decrease with temperature. Preferably, the temperature dependence of the characteristic of the temperature-dependent element 72 is approximately linear with temperature over the expected range of biological temperatures (about 25 to 45° C.), although this is not required.

Typically, a signal (e.g., a current) having an amplitude or other property that is a function of the temperature can be obtained by providing a potential across the two probe leads 68, 70 of the temperature probe 66. As the temperature changes, the temperature-dependent characteristic of the temperature-dependent element 72 increases or decreases with a corresponding change in the signal amplitude. The signal from the temperature probe 66 (e.g., the amount of current flowing through the probe) may be combined with the signal obtained from the working electrode 58 by, for example, scaling the temperature probe signal and then adding or subtracting the scaled temperature probe signal from the signal at the working electrode 58. In this manner, the temperature probe 66 can provide a temperature adjustment for the output from the working electrode 58 to offset the temperature dependence of the working electrode 58.

One embodiment of the temperature probe includes probe leads 68, 70 formed as two spaced-apart channels with a temperature-dependent element 72 formed as a cross-channel connecting the two spaced-apart channels, as illustrated in FIG. 8. The two spaced-apart channels contain a conductive material, such as a metal, alloy, semimetal, degenerate semiconductor, or metallic compound. The cross-channel may contain the same material (provided the cross-channel has a smaller cross-section than the two spaced-apart channels) as the probe leads 68, 70. In other

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embodiments, the material in the cross-channel is different than the material of the probe leads 68, 70.

One exemplary method for forming this particular temperature probe includes forming the two spaced-apart channels and then filling them with the metallic or alloyed conductive material. Next, the cross-channel is formed and then filled with the desired material. The material in the cross-channel overlaps with the conductive material in each of the two spaced-apart channels to form an electrical connection.

For proper operation of the temperature probe 66, the temperature-dependent element 72 of the temperature probe 66 can not be shorted by conductive material formed between the two probe leads 68, 70. In addition, to prevent conduction between the two probe leads 68, 70 by ionic species within the body or sample fluid, a covering may be provided over the temperature-dependent element 72, and preferably over the portion of the probe leads 68, 70 that is implanted in the patient. The covering may be, for example, a non-conducting film disposed over the temperature-dependent element 72 and probe leads 68, 70 to prevent the ionic conduction. Suitable non-conducting films include, for example, Kapton™ polyimide films (DuPont, Wilmington, Del.).

Another method for eliminating or reducing conduction by ionic species in the body or sample fluid is to use an ac voltage source connected to the probe leads 68, 70. In this way, the positive and negative ionic species are alternately attracted and repelled during each half cycle of the ac voltage. This results in no net attraction of the ions in the body or sample fluid to the temperature probe 66. The maximum amplitude of the ac current through the temperature-dependent element 72 may then be used to correct the measurements from the working electrodes 58.

The temperature probe can be placed on the same substrate as the electrodes. Alternatively, a temperature probe may be placed on a separate substrate. In addition, the temperature probe may be used by itself or in conjunction with other devices.

Another embodiment of a temperature probe utilizes the temperature dependence of the conductivity of a solution (e.g., blood or interstitial fluid). Typically, the conductivity of an electrolyte-containing solution is dependent on the temperature of the solution, assuming that the concentration of electrolytes is relatively constant. Blood, interstitial fluid, and other bodily fluids are solutions with relatively constant levels of electrolytes. Thus, a sensor 42 can include two or more conductive traces (not shown) which are spaced apart by a known distance. A portion of these conductive traces is exposed to the solution and the conductivity between the exposed portions of the conductive traces is measured using known techniques (e.g., application of a constant or known current or potential and measurement of the resulting potential or current, respectively, to determine the conductivity).

A change in conductivity is related to a change in temperature. This relation can be modeled using linear, quadratic, exponential, or other relations. The parameters for this relationship typically do not vary significantly between most people. The calibration for the temperature probe can be determined by a variety of methods, including, for example, calibration of each sensor 42 using an independent method of determining temperature (e.g., a thermometer, an optical or electrical temperature detector, or the temperature probe 66, described above) or calibrating one sensor 42 and using that calibration for all other sensors in a batch based on uniformity in geometry.

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Biocompatible Layer

An optional film layer **75** is formed over at least that portion of the sensor **42** which is subcutaneously inserted into the patient, as shown in FIG. **9**. This optional film layer **74** may serve one or more functions. The film layer **74** prevents the penetration of large biomolecules into the electrodes. This is accomplished by using a film layer **74** having a pore size that is smaller than the biomolecules that are to be excluded. Such biomolecules may foul the electrodes and/or the sensing layer **64** thereby reducing the effectiveness of the sensor **42** and altering the expected signal amplitude for a given analyte concentration. The fouling of the working electrodes **58** may also decrease the effective life of the sensor **42**. The biocompatible layer **74** may also prevent protein adhesion to the sensor **42**, formation of blood clots, and other undesirable interactions between the sensor **42** and body.

For example, the sensor may be completely or partially coated on its exterior with a biocompatible coating. A preferred biocompatible coating is a hydrogel which contains at least 20 wt. % fluid when in equilibrium with the analyte-containing fluid. Examples of suitable hydrogels are described in U.S. Pat. No. 5,593,852, incorporated herein by reference, and include crosslinked polyethylene oxides, such as polyethylene oxide tetraacrylate.

Interferent-Eliminating Layer

An interferent-eliminating layer (not shown) may be included in the sensor **42**. The interferent-eliminating layer may be incorporated in the biocompatible layer **75** or in the mass transport limiting layer **74** (described below) or may be a separate layer. Interferents are molecules or other species that are electroreduced or electrooxidized at the electrode, either directly or via an electron transfer agent, to produce a false signal. In one embodiment, a film or membrane prevents the penetration of one or more interferents into the region around the working electrodes **58**. Preferably, this type of interferent-eliminating layer is much less permeable to one or more of the interferents than to the analyte.

The interferent-eliminating layer may include ionic components, such as Nafion®, incorporated into a polymeric matrix to reduce the permeability of the interferent-eliminating layer to ionic interferents having the same charge as the ionic components. For example, negatively charged compounds or compounds that form negative ions may be incorporated in the interferent-eliminating layer to reduce the permeation of negative species in the body or sample fluid.

Another example of an interferent-eliminating layer includes a catalyst for catalyzing a reaction which removes interferents. One example of such a catalyst is a peroxidase. Hydrogen peroxide reacts with interferents, such as acetaminophen, urate, and ascorbate. The hydrogen peroxide may be added to the analyte-containing fluid or may be generated in situ, by, for example, the reaction of glucose or lactate in the presence of glucose oxidase or lactate oxidase, respectively. Examples of interferent eliminating layers include a peroxidase enzyme crosslinked (a) using gluteraldehyde as a crosslinking agent or (b) oxidation of oligosaccharide groups in the peroxidase glycoenzyme with $NaIO_4$, followed by coupling of the aldehydes formed to hydrazide groups in a polyacrylamide matrix to form hydrazones are describe in U.S. Pat. Nos. 5,262,305 and 5,356,786, incorporated herein by reference.

Mass Transport Limiting Layer

A mass transport limiting layer **74** may be included with the sensor to act as a diffusion-limiting barrier to reduce the rate of mass transport of the analyte, for example, glucose or lactate, into the region around the working electrodes **58**. By limiting the diffusion of the analyte, the steady state concentration of the analyte in the proximity of the working electrode **58** (which is proportional to the concentration of the analyte in the body or sample fluid) can be reduced. This extends the upper range of analyte concentrations that can still be accurately measured and may also expand the range in which the current increases approximately linearly with the level of the analyte.

It is preferred that the permeability of the analyte through the film layer **74** vary little or not at all with temperature, so as to reduce or eliminate the variation of current with temperature. For this reason, it is preferred that in the biologically relevant temperature range from about 25° C. to about 45° C., and most importantly from 30° C. to 40° C., neither the size of the pores in the film nor its hydration or swelling change excessively. Preferably, the mass transport limiting layer is made using a film that absorbs less than 5 wt. % of fluid over 24 hours. This may reduce or obviate any need for a temperature probe. For implantable sensors, it is preferable that the mass transport limiting layer is made using a film that absorbs less than 5 wt. % of fluid over 24 hours at 37° C.

Particularly useful materials for the film layer **74** are membranes that do not swell in the analyte-containing fluid that the sensor tests. Suitable membranes include 3 to 20,000 nm diameter pores. Membranes having 5 to 500 nm diameter pores with well-defined, uniform pore sizes and high aspect ratios are preferred. In one embodiment, the aspect ratio of the pores is preferably two or greater and more preferably five or greater.

Well-defined and uniform pores can be made by track etching a polymeric membrane using accelerated electrons, ions, or particles emitted by radioactive nuclei. Most preferred are anisotropic, polymeric, track etched membranes that expand less in the direction perpendicular to the pores than in the direction of the pores when heated. Suitable polymeric membranes included polycarbonate membranes from Poretics (Livermore, Calif., catalog number 19401, 0.01 μm pore size polycarbonate membrane) and Corning Costar Corp. (Cambridge, Mass., Nucleopore™ brand membranes with 0.015 μm pore size). Other polyolefin and polyester films may be used. It is preferred that the permeability of the mass transport limiting membrane changes no more than 4%, preferably, no more than 3%, and, more preferably, no more than 2%, per ° C. in the range from 30° C. to 40° C. when the membranes resides in the subcutaneous interstitial fluid.

In some embodiments of the invention, the mass transport limiting layer **74** may also limit the flow of oxygen into the sensor **42**. This can improve the stability of sensors **42** that are used in situations where variation in the partial pressure of oxygen causes non-linearity in sensor response. In these embodiments, the mass transport limiting layer **74** restricts oxygen transport by at least 40%, preferably at least 60%, and more preferably at least 80%, than the membrane restricts transport of the analyte. For a given type of polymer, films having a greater density (e.g., a density closer to that of the crystalline polymer) are preferred. Polyesters, such as polyethylene terephthalate, are typically less permeable to oxygen and are, therefore, preferred over polycarbonate membranes.

Anticlotting Agent

An implantable sensor may also, optionally, have an anticlotting agent disposed on a portion the substrate which is implanted into a patient. This anticlotting agent may reduce or eliminate the clotting of blood or other body fluid

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around the sensor, particularly after insertion of the sensor. Blood clots may foul the sensor or irreproducibly reduce the amount of analyte which diffuses into the sensor. Examples of useful anticlotting agents include heparin and tissue plasminogen activator (TPA), as well as other known anti-clotting agents.

The anticlotting agent may be applied to at least a portion of that part of the sensor **42** that is to be implanted. The anticlotting agent may be applied, for example, by bath, spraying, brushing, or dipping. The anticlotting agent is allowed to dry on the sensor **42**. The anticlotting agent may be immobilized on the surface of the sensor or it may be allowed to diffuse away from the sensor surface. Typically, the quantities of anticlotting agent disposed on the sensor are far below the amounts typically used for treatment of medical conditions involving blood clots and, therefore, have only a limited, localized effect.

Sensor Lifetime

The sensor **42** may be designed to be a replaceable component in an in vivo analyte monitor, and particularly in an implantable analyte monitor. Typically, the sensor **42** is capable of operation over a period of days. Preferably, the period of operation is at least one day, more preferably at least three days, and most preferably at least one week. The sensor **42** can then be removed and replaced with a new sensor. The lifetime of the sensor **42** may be reduced by the fouling of the electrodes or by the leaching of the electron transfer agent or catalyst. These limitations on the longevity of the sensor **42** can be overcome by the use of a biocom-patible layer **75** or non-leachable electron transfer agent and catalyst, respectively, as described above.

Another primary limitation on the lifetime of the sensor **42** is the temperature stability of the catalyst. Many catalysts are enzymes, which are very sensitive to the ambient tem-perature and may degrade at temperatures of the patient's body (e.g., approximately 37° C. for the human body). Thus, robust enzymes should be used where available. The sensor **42** should be replaced when a sufficient amount of the enzyme has been deactivated to introduce an unacceptable amount of error in the measurements.

Insertion Device

An insertion device **120** can be used to subcutaneously insert the sensor **42** into the patient, as illustrated in FIG. **12**. The insertion device **120** is typically formed using structur-ally rigid materials, such as metal or rigid plastic. Preferred materials include stainless steel and ABS (acrylonitrile-butadiene-styrene) plastic. In some embodiments, the inser-tion device **120** is pointed and/or sharp at the tip **121** to facilitate penetration of the skin of the patient. A sharp, thin insertion device may reduce pain felt by the patient upon insertion of the sensor **42**. In other embodiments, the tip **121** of the insertion device **120** has other shapes, including a blunt or flat shape. These embodiments may be particularly useful when the insertion device **120** does not penetrate the skin but rather serves as a structural support for the sensor **42** as the sensor **42** is pushed into the skin.

The insertion device **120** may have a variety of cross-sectional shapes, as shown in FIGS. **13A**, **13B**, and **13C**. The insertion device **120** illustrated in FIG. **13A** is a flat, planar, pointed strip of rigid material which may be attached or otherwise coupled to the sensor **42** to ease insertion of the sensor **42** into the skin of the patient, as well as to provide structural support to the sensor **42** during insertion. The insertion devices **120** of FIGS. **13B** and **13C** are U- or V-shaped implements that support the sensor **42** to limit the amount that the sensor **42** may bend or bow during insertion. The cross-sectional width **124** of the insertion devices **120**

illustrated in FIGS. **13B** and **13C** is typically 1 mm or less, preferably 700 μm or less, more preferably 500 μm or less, and most preferably 300 μm or less. The cross-sectional height **126** of the insertion device **120** illustrated in FIGS. **13B** and **13C** is typically about 1 mm or less, preferably about 700 μm or less, and more preferably about 500 μm or less.

The sensor **42** itself may include optional features to facilitate insertion. For example, the sensor **42** may be pointed at the tip **123** to ease insertion, as illustrated in FIG. **12**. In addition, the sensor **42** may include a barb **125** which helps retain the sensor **42** in the subcutaneous tissue of the patient. The barb **125** may also assist in anchoring the sensor **42** within the subcutaneous tissue of the patient during operation of the sensor **42**. However, the barb **125** is typically small enough that little damage is caused to the subcutaneous tissue when the sensor **42** is removed for replacement. The sensor **42** may also include a notch **127** that can be used in cooperation with a corresponding struc-ture (not shown) in the insertion device to apply pressure against the sensor **42** during insertion, but disengage as the insertion device **120** is removed. One example of such a structure in the insertion device is a rod (not shown) between two opposing sides of an insertion device **120** and at an appropriate height of the insertion device **120**.

In operation, the sensor **42** is placed within or next to the insertion device **120** and then a force is provided against the insertion device **120** and/or sensor **42** to carry the sensor **42** into the skin of the patient. In one embodiment, the force is applied to the sensor **42** to push the sensor into the skin, while the insertion device **120** remains stationary and pro-vides structural support to the sensor **42**. Alternatively, the force is applied to the insertion device **120** and optionally to the sensor **42** to push a portion of both the sensor **42** and the insertion device **120** through the skin of the patient and into the subcutaneous tissue. The insertion device **120** is option-ally pulled out of the skin and subcutaneous tissue with the sensor **42** remaining in the subcutaneous tissue due to frictional forces between the sensor **42** and the patient's tissue. If the sensor **42** includes the optional barb **125**, then this structure may also facilitate the retention of the sensor **42** within the interstitial tissue as the barb catches in the tissue.

The force applied to the insertion device **120** and/or the sensor **42** may be applied manually or mechanically. Preferably, the sensor **42** is reproducibly inserted through the skin of the patient. In one embodiment, an insertion gun is used to insert the sensor. One example of an insertion gun **200** for inserting a sensor **42** is shown in FIG. **26**. The insertion gun **200** includes a housing **202** and a carrier **204**. The insertion device **120** is typically mounted on the carrier **204** and the sensor **42** is pre-loaded into the insertion device **120**. The carrier **204** drives the sensor **42** and, optionally, the insertion device **120** into the skin of the patient using, for example, a cocked or wound spring, a burst of compressed gas, an electromagnet repelled by a second magnet, or the like, within the insertion gun **200**. In some instances, for example, when using a spring, the carrier **204** and insertion device may be moved, cocked, or otherwise prepared to be directed towards the skin of the patient.

After the sensor **42** is inserted, the insertion gun **200** may contain a mechanism which pulls the insertion device **120** out of the skin of the patient. Such a mechanism may use a spring, electromagnet, or the like to remove the insertion device **120**.

The insertion gun may be reusable. The insertion device **120** is often disposable to avoid the possibility of contami-

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nation. Alternatively, the insertion device **120** may be sterilized and reused. In addition, the insertion device **120** and/or the sensor **42** may be coated with an anticlotting agent to prevent fouling of the sensor **42**.

In one embodiment, the sensor **42** is injected between 2 to 12 mm into the interstitial tissue of the patient for subcutaneous implantation. Preferably, the sensor is injected 3 to 9 mm, and more preferably 5 to 7 mm, into the interstitial tissue. Other embodiments of the invention, may include sensors implanted in other portions of the patient, including, for example, in an artery, vein, or organ. The depth of implantation varies depending on the desired implantation target.

Although the sensor **42** may be inserted anywhere in the body, it is often desirable that the insertion site be positioned so that the on-skin sensor control unit **44** can be concealed. In addition, it is often desirable that the insertion site be at a place on the body with a low density of nerve endings to reduce the pain to the patient. Examples of preferred sites for insertion of the sensor **42** and positioning of the on-skin sensor control unit **44** include the abdomen, thigh, leg, upper arm, and shoulder.

An insertion angle is measured from the plane of the skin (i.e., inserting the sensor perpendicular to the skin would be a 90° insertion angle). Insertion angles usually range from 10 to 90°, typically from 15 to 60°, and often from 30 to 45°.

On-Skin Sensor Control Unit

The on-skin sensor control unit **44** is configured to be placed on the skin of a patient. The on-skin sensor control unit **44** is optionally formed in a shape that is comfortable to the patient and which may permit concealment, for example, under a patient's clothing. The thigh, leg, upper arm, shoulder, or abdomen are convenient parts of the patient's body for placement of the on-skin sensor control unit **44** to maintain concealment. However, the on-skin sensor control unit **44** may be positioned on other portions of the patient's body. One embodiment of the on-skin sensor control unit **44** has a thin, oval shape to enhance concealment, as illustrated in FIGS. **14–16**. However, other shapes and sizes may be used.

The particular profile, as well as the height, width, length, weight, and volume of the on-skin sensor control unit **44** may vary and depends, at least in part, on the components and associated functions included in the on-skin sensor control unit **44**, as discussed below. For example, in some embodiments, the on-skin sensor control unit **44** has a height of 1.3 cm or less, and preferably 0.7 cm or less. In some embodiments, the on-skin sensor control unit **44** has a weight of 90 grams or less, preferably 45 grams or less, and more preferably 25 grams or less. In some embodiments, the on-skin sensor control unit **44** has a volume of about 15 cm^3 or less, preferably about 10 cm^3 or less, more preferably about 5 cm^3 or less, and most preferably about 2.5 cm^3 or less.

The on-skin sensor control unit **44** includes a housing **45**, as illustrated in FIGS. **14–16**. The housing **45** is typically formed as a single integral unit that rests on the skin of the patient. The housing **45** typically contains most or all of the electronic components, described below, of the on-skin sensor control unit **44**. The on-skin sensor control unit **44** usually includes no additional cables or wires to other electronic components or other devices. If the housing includes two or more parts, then those parts typically fit together to form a single integral unit.

The housing **45** of the on-skin sensor control unit **44**, illustrated in FIGS. **14–16**, may be formed using a variety of materials, including, for example, plastic and polymeric

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materials, particularly rigid thermoplastics and engineering thermoplastics. Suitable materials include, for example, polyvinyl chloride, polyethylene, polypropylene, polystyrene, ABS polymers, and copolymers thereof. The housing **45** of the on-skin sensor control unit **44** may be formed using a variety of techniques including, for example, injection molding, compression molding, casting, and other molding methods. Hollow or recessed regions may be formed in the housing **45** of the on-skin sensor control unit **44**. The electronic components of the on-skin sensor control unit **44**, described below, and/or other items, such as a battery or a speaker for an audible alarm, may be placed in the hollow or recessed areas.

In some embodiments, conductive contacts **80** are provided on the exterior of the housing **45**. In other embodiments, the conductive contacts **80** are provided on the interior of the housing **45**, for example, within a hollow or recessed region.

In some embodiments, the electronic components and/or other items are incorporated into the housing **45** of the on-skin sensor control unit **44** as the plastic or polymeric material is molded or otherwise formed. In other embodiments, the electronic components and/or other items are incorporated into the housing **45** as the molded material is cooling or after the molded material has been reheated to make it pliable. Alternatively, the electronic components and/or other items may be secured to the housing **45** using fasteners, such as screws, nuts and bolts, nails, staples, rivets, and the like or adhesives, such as contact adhesives, pressure sensitive adhesives, glues, epoxies, adhesive resins, and the like. In some cases, the electronic components and/or other items are not affixed to the housing **45** at all.

In some embodiments, the housing **45** of the on-skin sensor control unit **44** is a single piece. The conductive contacts **80** may be formed on the exterior of the housing **45** or on the interior of the housing **45** provided there is a port **78** in the housing **45** through which the sensor **42** can be directed to access the conductive contacts **80**.

In other embodiments, the housing **45** of the on-skin sensor control unit **44** is formed in at least two separate portions that fit together to form the housing **45**, for example, a base **74** and a cover **76**, as illustrated in FIGS. **14–16**. The two or more portions of the housing **45** may be entirely separate from each other. Alternatively, at least some of the two or more portions of the housing **45** may be connected together, for example, by a hinge, to facilitate the coupling of the portions to form the housing **45** of the on-skin sensor control unit **44**.

These two or more separate portions of the housing **45** of on-skin sensor control unit **44** may have complementary, interlocking structures, such as, for example, interlocking ridges or a ridge on one component and a complementary groove on another component, so that the two or more separate components may be easily and/or firmly coupled together. This may be useful, particularly if the components are taken apart and fit together occasionally, for example, when a battery or sensor **42** is replaced. However, other fasteners may also be used to couple the two or more components together, including, for example, screws, nuts and bolts, nails, staples, rivets, or the like. In addition, adhesives, both permanent or temporary, may be used including, for example, contact adhesives, pressure sensitive adhesives, glues, epoxies, adhesive resins, and the like.

Typically, the housing **45** is at least water resistant to prevent the flow of fluids into contact with the components in the housing, including, for example, the conductive contacts **80**. Preferably, the housing is waterproof. In one

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embodiment, two or more components of the housing **45**, for example, the base **74** and the cover **76**, fit together tightly to form a hermetic, waterproof, or water resistant seal so that fluids can not flow into the interior of the on-skin sensor control unit **44**. This may be useful to avoid corrosion currents and/or degradation of items within the on-skin sensor control unit **44**, such as the conductive contacts, the battery, or the electronic components, particularly when the patient engages in such activities as showering, bathing, or swimming.

Water resistant, as used herein, means that there is no penetration of water through a water resistant seal or housing when immersed in water at a depth of one meter at sea level. Waterproof, as used herein, means that there is no penetration of water through the waterproof seal or housing when immersed in water at a depth of ten meters, and preferably fifty meters, at sea level. It is often desirable that the electronic circuitry, power supply (e.g., battery), and conductive contacts of the on-skin sensor control unit, as well as the contact pads of the sensor, are contained in a water resistant, and preferably, a waterproof, environment.

In addition to the portions of the housing **45**, such as the base **74** and cover **76**, there may be other individually-formed pieces of the on-skin sensor control unit **44**, which may be assembled during or after manufacture. One example of an individually-formed piece is a cover for electronic components that fits a recess in the base **74** or cover **76**. Another example is a cover for a battery provided in the base **74** or cover **76**. These individually-formed pieces of the on-skin sensor control unit **44** may be permanently affixed, such as, for example, a cover for electronic components, or removably affixed, such as, for example, a removable cover for a battery, to the base **74**, cover **76**, or other component of the on-skin sensor control unit **44**. Methods for affixing these individually-formed pieces include the use of fasteners, such as screws, nuts and bolts, staples, nails, rivets, and the like, frictional fasteners, such as tongue and groove structures, and adhesives, such as contact adhesives, pressure sensitive adhesives, glues, epoxies, adhesive resins, and the like.

One embodiment of the on-skin sensor control unit **44** is a disposable unit complete with a battery for operating the unit. There are no portions of the unit that the patient needs to open or remove, thereby reducing the size of the unit and simplifying its construction. The on-skin sensor control unit **44** optionally remains in a sleep mode prior to use to conserve the battery's power. The on-skin sensor control unit **44** detects that it is being used and activates itself. Detection of use may be through a number of mechanisms. These include, for example, detection of a change in resistance across the electrical contacts, actuation of a switch upon mating the on-skin sensor control unit **44** with a mounting unit **77** (see FIGS. **27A** and **28A**). The on-skin sensor control unit **44** is typically replaced when it no longer operates within threshold limits, for example, if the battery or other power source does not generate sufficient power. Often this embodiment of the on-skin sensor control unit **44** has conductive contacts **80** on the exterior of the housing **45**. Once the sensor **42** is implanted in the patient, the sensor control unit **44** is placed over the sensor **42** with the conductive contacts **80** in contact with the contact pads **49** of the sensor **42**.

The on-skin sensor control unit **44** is typically attached to the skin **75** of the patient, as illustrated in FIG. **17**. The on-skin sensor control unit **44** may be attached by a variety of techniques including, for example, by adhering the on-skin sensor control unit **44** directly to the skin **75** of the

patient with an adhesive provided on at least a portion of the housing **45** of the on-skin sensor control unit **44** which contacts the skin **75** or by suturing the on-skin sensor control unit **44** to the skin **75** through suture openings (not shown) in the sensor control unit **44**.

Another method of attaching the housing **45** of the on-skin sensor control unit **44** to the skin **75** includes using a mounting unit, **77**. The mounting unit **77** is often a part of the on-skin sensor control unit **44**. One example of a suitable mounting unit **77** is a double-sided adhesive strip, one side of which is adhered to a surface of the skin of the patient and the other side is adhered to the on-skin sensor control unit **44**. In this embodiment, the mounting unit **77** may have an optional opening **79** which is large enough to allow insertion of the sensor **42** through the opening **79**. Alternatively, the sensor may be inserted through a thin adhesive and into the skin.

A variety of adhesives may be used to adhere the on-skin sensor control unit **44** to the skin **75** of the patient, either directly or using the mounting unit **77**, including, for example, pressure sensitive adhesives (PSA) or contact adhesives. Preferably, an adhesive is chosen which is not irritating to all or a majority of patients for at least the period of time that a particular sensor **42** is implanted in the patient. Alternatively, a second adhesive or other skin-protecting compound may be included with the mounting unit so that a patient, whose skin is irritated by the adhesive on the mounting unit **77**, can cover his skin with the second adhesive or other skin-protecting compound and then place the mounting unit **77** over the second adhesive or other skin-protecting compound. This should substantially prevent the irritation of the skin of the patient because the adhesive on the mounting unit **77** is no longer in contact with the skin, but is instead in contact with the second adhesive or other skin-protecting compound.

When the sensor **42** is changed, the on-skin sensor control unit **44** may be moved to a different position on the skin **75** of the patient, for example, to avoid excessive irritation. Alternatively, the on-skin sensor control unit **44** may remain at the same place on the skin of the patient until it is determined that the unit **44** should be moved.

Another embodiment of a mounting unit **77** used in an on-skin sensor control unit **44** is illustrated in FIGS. **27A** and **27B**. The mounting unit **77** and a housing **45** of an on-skin sensor control unit **44** are mounted together in, for example, an interlocking manner, as shown in FIG. **27A**. The mounting unit **77** is formed, for example, using plastic or polymer materials, including, for example, polyvinyl chloride, polyethylene, polypropylene, polystyrene, ABS polymers, and copolymers thereof. The mounting unit **77** may be formed using a variety of techniques including, for example, injection molding, compression molding, casting, and other molding methods.

The mounting unit **77** typically includes an adhesive on a bottom surface of the mounting unit **77** to adhere to the skin of the patient or the mounting unit **77** is used in conjunction with, for example, double-sided adhesive tape or the like. The mounting unit **77** typically includes an opening **79** through which the sensor **42** is inserted, as shown in FIG. **27B**. The mounting unit **77** may also include a support structure **220** for holding the sensor **42** in place and against the conductive contacts **80** on the on-skin sensor control unit **42**. The mounting unit **77**, also, optionally, includes a positioning structure **222**, such as an extension of material from the mounting unit **77**, that corresponds to a structure (not shown), such as an opening, on the sensor **42** to facilitate proper positioning of the sensor **42**, for example, by aligning the two complementary structures.

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In another embodiment, a coupled mounting unit **77** and housing **45** of an on-skin sensor control unit **44** is provided on an adhesive patch **204** with an optional cover **206** to protect and/or confine the housing **45** of the on-skin sensor control unit **44**, as illustrated in FIG. **28A**. The optional cover may contain an adhesive or other mechanism for attachment to the housing **45** and/or mounting unit **77**. The mounting unit **77** typically includes an opening **49** through which a sensor **42** is disposed, as shown in FIG. **28B**. The opening **49** may optionally be configured to allow insertion of the sensor **42** through the opening **49** using an insertion device **120** or insertion gun **200** (see FIG. **26**). The housing **45** of the on-skin sensor control unit **44** has a base **74** and a cover **76**, as illustrated in FIG. **28C**. A bottom view of the housing **45**, as shown in FIG. **28D**, illustrates ports **230** through which conductive contacts (not shown) extend to connect with contact pads on the sensor **42**. A board **232** for attachment of circuit components may optionally be provided within the on-skin sensor control unit **44**, as illustrated in FIG. **28E**.

In some embodiments, the adhesive on the on-skin sensor control unit **44** and/or on any of the embodiments of the mounting unit **77** is water resistant or waterproof to permit activities such as showering and/or bathing while maintaining adherence of the on-skin sensor control unit **44** to the skin **75** of the patient and, at least in some embodiments, preventing water from penetrating into the sensor control unit **44**. The use of a water resistant or waterproof adhesive combined with a water resistant or waterproof housing **45** protects the components in the sensor control unit **44** and the contact between the conductive contacts **80** and the sensor **42** from damage or corrosion. An example of a non-irritating adhesive that repels water is Tegaderm (3M, St. Paul, Minn.).

In one embodiment, the on-skin sensor control unit **44** includes a sensor port **78** through which the sensor **42** enters the subcutaneous tissue of the patient, as shown in FIGS. **14** to **16**. The sensor **42** may be inserted into the subcutaneous tissue of the patient through the sensor port **78**. The on-skin sensor control unit **44** may then be placed on the skin of the patient with the sensor **42** being threaded through the sensor port **78**. If the housing **45** of the sensor **42** has, for example, a base **74** and a cover **76**, then the cover **76** may be removed to allow the patient to guide the sensor **42** into the proper position for contact with the conductive contacts **80**.

Alternatively, if the conductive contacts **80** are within the housing **45** the patient may slide the sensor **42** into the housing **45** until contact is made between the contact pads **49** and the conductive contacts **80**. The sensor control unit **44** may have a structure which obstructs the sliding of the sensor **42** further into the housing once the sensor **42** is properly positioned with the contact pads **49** in contact with the conductive contacts **80**.

In other embodiments, the conductive contacts **80** are on the exterior of the housing **45** (see e.g., FIGS. **27A–27B** and **28A–28E**). In these embodiments, the patient guides the contacts pads **49** of the sensor **42** into contact with the conductive contacts **80**. In some cases, a guiding structure may be provided on the housing **45** which guides the sensor **42** into the proper position. An example of such a structure includes a set of guiding rails extending from the housing **45** and having the shape of the sensor **42**.

In some embodiments, when the sensor **42** is inserted using an insertion device **120** (see FIG. **12**), the tip of the insertion device **120** or optional insertion gun **200** (see FIG. **26**) is positioned against the skin or the mounting unit **77** at the desired insertion point. In some embodiments, the inser-

tion device **120** is positioned on the skin without any guide. In other embodiments, the insertion device **120** or insertion gun **200** is positioned using guides (not shown) in the mounting unit **77** or other portion of the on-skin sensor control unit **44**. In some embodiments, the guides, opening **79** in the mounting unit **77** and/or sensor port **78** in the housing **45** of the on-skin sensor control unit **44** have a shape which is complementary to the shape of the tip of the insertion device **120** and/or insertion gun **200** to limit the orientation of the insertion device **120** and/or insertion gun **200** relative to the opening **79** and/or sensor port **78**. The sensor can then be subcutaneously inserted into the patient by matching the complementary shape of the opening **79** or sensor port **78** with the insertion device **120** and/or insertion gun **200**.

In some embodiments, the shapes of a) the guides, opening **79**, or sensor port **78**, and (b) the insertion device **120** or insertion gun **200** are configured such that the two shapes can only be matched in a single orientation. This aids in inserting the sensor **42** in the same orientation each time a new sensor is inserted into the patient. This uniformity in insertion orientation may be required in some embodiments to ensure that the contact pads **49** on the sensor **42** are correctly aligned with appropriate conductive contacts **80** on the on-skin sensor control unit **44**. In addition, the use of the insertion gun, as described above, may ensure that the sensor **42** is inserted at a uniform, reproducible depth.

The sensor **42** and the electronic components within the on-skin sensor control unit **44** are coupled via conductive contacts **80**, as shown in FIGS. **14–16**. The one or more working electrodes **58**, counter electrode **60** (or counter/reference electrode), optional reference electrode **62**, and optional temperature probe **66** are attached to individual conductive contacts **80**. In the illustrated embodiment of FIGS. **14–16**, the conductive contacts **80** are provided on the interior of the on-skin sensor control unit **44**. Other embodiments of the on-skin sensor control unit **44** have the conductive contacts disposed on the exterior of the housing **45**. The placement of the conductive contacts **80** is such that they are in contact with the contact pads **49** on the sensor **42** when the sensor **42** is properly positioned within the on-skin sensor control unit **44**.

In the illustrated embodiment of FIGS. **14–16**, the base **74** and cover **76** of the on-skin sensor control unit **44** are formed such that, when the sensor **42** is within the on-skin sensor control unit **44** and the base **74** and cover **76** are fitted together, the sensor **42** is bent. In this manner, the contact pads **49** on the sensor **42** are brought into contact with the conductive contacts **80** of the on-skin sensor control unit **44**. The on-skin sensor control unit **44** may optionally contain a support structure **82** to hold, support, and/or guide the sensor **42** into the correct position.

Non-limiting examples of suitable conductive contacts **80** are illustrated in FIGS. **19A–19D**. In one embodiment, the conductive contacts **80** are pins **84** or the like, as illustrated in FIG. **19A**, which are brought into contact with the contact pads **49** on the sensor **42** when the components of the on-skin sensor control unit **44**, for example, the base **74** and cover **76**, are fitted together. A support **82** may be provided under the sensor **42** to promote adequate contact between the contact pads **49** on the sensor **42** and the pins **84**. The pins are typically made using a conductive material, such as a metal or alloy, for example, copper, stainless steel, or silver. Each pin has a distal end that extends from the on-skin sensor control unit **44** for contacting the contact pads **49** on the sensor **42**. Each pin **84** also has a proximal end that is coupled to a wire or other conductive strip that is, in turn,